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Product name

SECTION 1: Product and company identification

: Pellet Deicer



Use of the substance/mixture	: Ice melter
Product code	: 0600
Company	: Total Solutions P.O. Box 240014 Milwaukee, WI 53224 - USA T (414) 354-6417
Emergency number	: Chemtrec: (800) 424-9300
SECTION 2: Hazards iden	itification
2.1. Classification of the subs	tance or mixture
GHS-US classification	
Acute Tox. 4 (Oral) H302 Skin Irrit. 2 H315 Eye Irrit. 2B H320	
2.2. Label elements	
GHS-US labeling	
Hazard pictograms (GHS-US)	GHS07
Signal word (GHS-US)	: Warning
Hazard statements (GHS-US)	: Harmful if swallowed Causes skin irritation Causes eye irritation
Precautionary statements (GHS	S-US) : Wash thoroughly after handling Do not eat, drink or smoke when using this product. Wear eye protection, face protection, protective gloves. If swallowed: Call a doctor, a POISON CENTER if you feel unwell If on skin: Wash with plenty of water If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing Specific treatment (see First aid measures on this label) Rinse mouth. If skin irritation occurs: Get medical advice/attention. If eye irritation persists: Get medical advice/attention. Take off contaminated clothing and wash it before reuse. Dispose of contents/container to comply with local/regional/national/international regulations
2.3. Other hazards	

2.3. Other hazards No additional information available

2.4. Unknown acute toxicity (GHS US)

Not applicable

SECTION 3: Composition/Information on ingredients

3.1. Substances

Not applicable

Full text of H-phrases: see section 16

3.2. Mixtures

5.2. Mixtures			
Name	Product identifier	%	GHS-US classification
calcium chloride	(CAS-No.) 10043-52-4	> 90	Eye Irrit. 2A, H319
AQUA	(CAS-No.) 7732-18-5	< 10	Not classified
Potassium Chloride	(CAS-No.) 7447-40-7	< 3	Not classified
SODIUM CHLORIDE	(CAS-No.) 7647-14-5	< 2	Not classified
	of composition has been withheld as a trade secret. Any	. –	

A specific chemical identity and/or percentage of composition has been withheld as a trade secret. Any concentration shown as a range is to protect confidentiality or is due to batch variation.

SECTION 4: First aid measures

4.1. Description of first aid measures	
First-aid measures general	: If you feel unwell, seek medical advice (show the label where possible).



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First-aid measures after inhalation	: Remove person to fresh air and keep comfortable for breathing. Call a poison center/doctor/physician if you feel unwell.
irst-aid measures after skin contact	: Rinse skin with water/shower.
First-aid measures after eye contact	 Rinse immediately with plenty of water for 15 minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a poison center or doctor/physician.
First-aid measures after ingestion	: Victim is fully conscious: immediately induce vomiting. Immediately call a poison center or doctor/physician.
4.2. Most important symptoms and ef	
Symptoms/effects	: Harmful if swallowed. Causes eye irritation. Causes skin irritation.
Symptoms/effects after skin contact	Causes skin irritation.
Symptoms/effects after eye contact	: Causes eye irritation.
Symptoms/effects after ingestion	: Harmful if swallowed.
	ical attention and special treatment needed treat as any thermal burn, after decontamination. No specific antidote. Supportive care. Treatment based on reaction of the patient.
SECTION 5: Firefighting measur	es
5.1. Extinguishing media	
Suitable extinguishing media	: Adapt extinguishing media to the environment.
5.2. Special hazards arising from the s	
Fire hazard	 Not flammable. Heat and acid contamination will produce irritating and toxic fumes. May decompose, generating irritating chlorine gas. Heat is generated when mixed with water or aqueous acids. Spattering and boiling can occur. Avoid contact with: bromide trifluoride, 2-furan percarboxylic acid because calcium chloride is incompatible with those substances. Contact with zinc forms flammable hydrogen gas, which can be explosive. Catalyzes exothermic polymerization of methyl vinyl ether. Attacks metals in the presence of moisture, and may release flammable hydrogen gas. Reaction of bromide impurity with oxidizing materials may generate trace levels of impurities such as bromates.
Reactivity	: Contact with metallic substances may release flammable hydrogen gas. Thermal decomposition may produce chlorine, sodium oxide, oxygen, oxides of chlorine, sodium chlorate, and hydrogen. HCl.
5.3. Advice for firefighters	
Firefighting instructions	: Exercise caution when fighting any chemical fire.
Protection during firefighting	: Do not enter fire area without proper protective equipment, including respiratory protection.
SECTION 6: Accidental release n	ieasures
6.1. Personal precautions, protective	
General measures	: Isolate from fire, if possible, without unnecessary risk.
6.1.1. For non-emergency personnel	
Protective equipment	: Gloves. Safety glasses. Protective clothing.
Emergency procedures	: Avoid contact with eyes. Wash contaminated clothes.
6.1.2. For emergency responders	
Protective equipment	: Do not attempt to take action without suitable protective equipment.
Emergency procedures	: Stop leak if safe to do so. Stop release. Ventilate area.
6.2. Environmental precautions Avoid release to the environment.	
6.3. Methods and material for contain	
For containment	: Collect spillage.
Methods for cleaning up	: Clean contaminated surfaces with an excess of water. On land, sweep or shovel into suitable containers. This material and its container must be disposed of in a safe way, and as per local legislation.
6.4. Reference to other sections No additional information available	
SECTION 7: Handling and storag	
7.1. Precautions for safe handling	
Additional hazards when processed	: Always use cool water (temperature less than 80F, 27C) when dissolving calcium chloride. Heat developed by solution is very high during dissolving and mixing.
Precautions for safe handling	: Avoid contact with eyes.

Precautions for safe handling : Avoid contact with eyes. : Wash hands and other exposed areas with mild soap and water before eating, drinking or

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smoking and when leaving work.

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7.2. Conditions for safe storage, includ	ing any incompatibilities
Technical measures	: Comply with applicable regulations.
Storage conditions	: Keep container tightly closed.

- : Most the log
- Storage area Special rules on packaging

: Meet the legal requirements.

: Keep only in original container.

SECTION 8: Exposure controls/personal protection

calcium chloride (10043-52-4)		
Not applicable		
SODIUM CHLORIDE (7647-14-5)		
Not applicable		
Potassium Chloride (7447-40-7)		
Not applicable		
AQUA (7732-18-5)		
Not applicable		

8.2. Exposure controls

Appropriate engineering controls Personal protective equipment

SECTION 9: Physical and chemical properties

- : Ensure good ventilation of the work station.
- : Gloves. Safety glasses. Protective clothing. Use appropriate personal protective equipment when risk assessment indicates this is necessary.



Physical state	: Solid
Appearance	: White solid.
Odor	: No odor
Odor threshold	: No data available
рН	: No data available
Melting point	: No data available
Freezing point	: No data available
Boiling point	: No data available
Flash point	: No data available
Relative evaporation rate (butyl acetate=1)	: No data available
Flammability (solid, gas)	: No data available
Explosion limits	: No data available
Explosive properties	: No data available
Oxidizing properties	: No data available
Vapor pressure	: No data available
Relative density	: No data available
Relative vapor density at 20 °C	: No data available
Specific gravity / density	: 2.2 g/ml
Solubility	: Soluble in water.
Log Pow	: No data available
Log Kow	: No data available
Auto-ignition temperature	: No data available
Decomposition temperature	: No data available
Viscosity	: No data available
Viscosity, kinematic	: No data available
Viscosity, dynamic	: No data available

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SECTION 10: Stability and reactivity

10.1. Reactivity

Contact with metallic substances may release flammable hydrogen gas. Thermal decomposition may produce chlorine, sodium oxide, oxygen, oxides of chlorine, sodium chlorate, and hydrogen. HCl.

10.2. Chemical stability Hygroscopic. 10.3. Possibility of hazardous reactions No additional information available 10.4. Conditions to avoid No additional information available 10.5. Incompatible materials May be corrosive to metals. 10.6. Hazardous decomposition products Under normal conditions of storage and use, hazardous decomposition products should not be produced. **SECTION 11: Toxicological information** 11.1. Information on toxicological effects Acute toxicity : Oral: Harmful if swallowed. Skin corrosion/irritation Causes skin irritation. Serious eye damage/irritation Causes eye irritation. • Respiratory or skin sensitization Not classified Germ cell mutagenicity Not classified Carcinogenicity Not classified Reproductive toxicity ÷ Not classified Specific target organ toxicity - single exposure : Not classified Specific target organ toxicity - repeated Not classified exposure Aspiration hazard : Not classified Symptoms/effects after skin contact : Causes skin irritation. Symptoms/effects after eye contact : Causes eye irritation. Symptoms/effects after ingestion : Harmful if swallowed.

Likely routes of exposure	: Skin and eye contact
SECTION 12: Ecological information	on
12.1. Toxicity	
No additional information available	
12.2. Persistence and degradability	
No additional information available	
12.3. Bioaccumulative potential	
No additional information available	
SECTION 13: Disposal considerati	ons
13.1. Waste treatment methods	
Waste treatment methods	: Dispose of contents/container to comply with local/regional/national regulations.
SECTION 14: Transport informati	on
Department of Transportation (DOT)	
In accordance with DOT : Not regulated	d for transport
Additional information	
Other information	: No supplementary information available.
ADR	
No additional information available	
Transport by sea	
No additional information available	
Air transport	

No additional information available

SECTION 15: Regulatory information

All components of this product are listed, or excluded from listing, on the United States Environmental Protection Agency Toxic Substances Control Act (TSCA) inventory

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This product or mixture does not contain a toxic chemical or chemicals in excess of the applicable de minimis concentration as specified in 40 CFR §372.38(a) subject to the reporting requirements of section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR Part 372.

This product contains no listed substances known to the State of California to cause cancer, birth defects or other reproductive harm, at levels which would require a warning under the statute. **WARNING** This product (when used in aqueous formulations with a chemical oxidizer such as ozone) may react to form calcium bromate, a chemical known to the State of California to cause cancer.

Training advice	: Normal use of this product shall imply use in accordance with the instructions on the packaging.
Full text of H-phrase	
H319	Causes serious eye irritation
NFPA health hazard NFPA fire hazard	 1 - Materials that, under emergency conditions, can cause significant irritation. 0 - Materials that will not burn under typical fire conditions, including intrinsically noncombustible materials such as concrete, stone, and sand.
NFPA reactivity	: 0 - Material that in themselves are normally stable, even under fire conditions.

Prepared by: Technical Department

This information is based on our current knowledge and is intended to describe the product for the purposes of health, safety and environmental requirements only. It should not therefore be construed as guaranteeing any specific property of the product. No warranty is expressed or implied regarding the accuracy of this data or the results obtained from the use thereof. Our company assumes no responsibility for personal injury or property damage to the vendee, users or third parties caused by the material. Such vendees or users assume all risks associated with the use of this material.